

## A Facile Synthesis of Deoxycorticosterone using the Controlled Alkaline Hydrolysis of 21-Bromo-20-ketopregnenes

Mitsuteru Numazawa\* and Masao Nagaoka

Tohoku College of Pharmacy, Komatsushima, Sendai, Miyagi 983, Japan

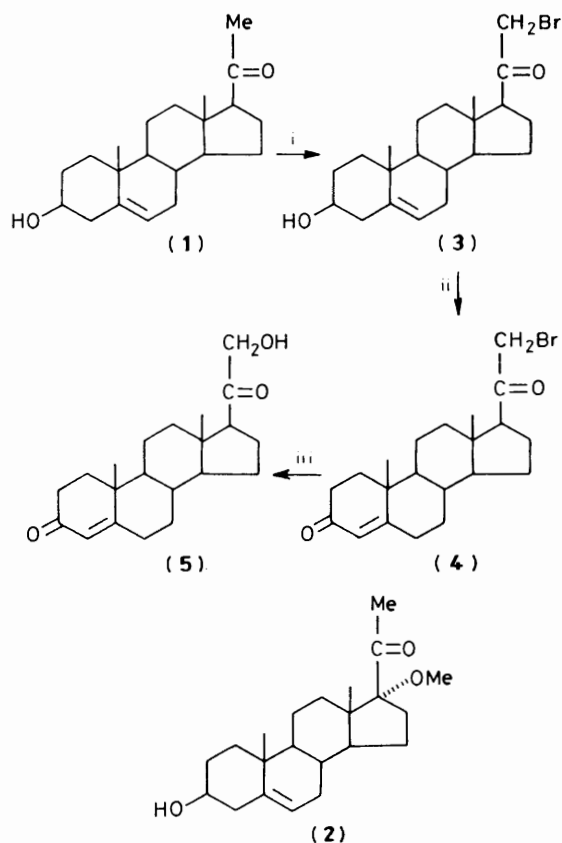
A novel synthesis of deoxycorticosterone (**5**) has been accomplished using the controlled alkaline hydrolysis of 21-bromopregn-4-ene-3,20-dione (**4**), obtained by the direct bromination of 3 $\beta$ -hydroxypregn-5-en-20-one (**1**) with CuBr<sub>2</sub> in the presence of pyridine, followed by oxidation with 2.67 M CrO<sub>3</sub> and subsequent acid-treatment.

Introduction of the 21-oxygen function into the 17 $\beta$ -acetyl side-chain of pregnenes is of central importance in the partial synthesis of corticoids. Probably the most widely used method for this synthesis involves the halogenation of 20-ketopregnenes, followed by displacement of 21-halides by acetate.<sup>1</sup> From observations on the bromination of 3 $\beta$ -hydroxypregn-5-en-20-one (**1**) with common reagents (*e.g.*, Br<sub>2</sub><sup>1</sup> or CuBr<sub>2</sub><sup>2</sup>), it has been considered impossible to obtain the 21-bromide (**3**) from the 20-ketone (**1**) directly. Moreover, efficient displacement of the bromo-group of a 21-bromo-20-ketone by a hydroxy-group has not been achieved probably owing to the Favorskii rearrangement<sup>3</sup> involved in the reaction and the sensitivity of the product, 21-hydroxy-20-ketone, to basic reagents. We now report a novel synthesis of deoxycorticosterone (**5**) which involves the direct bromination at C-21 of the 20-ketone (**1**) and the subsequent alkaline hydrolysis of the 21-bromide (**3**) under the controlled conditions<sup>4</sup> which we recently discovered and utilized in the synthesis of 16 $\alpha$ -hydroxy-17-keto- and 2 $\alpha$ -hydroxy-3-keto-steroids.

Treatment of 20-ketopregnene (**1**) with CuBr<sub>2</sub> (3 mol. equiv.,<sup>4c</sup> dry MeOH, reflux, 24 h) gave 3 $\beta$ -hydroxy-17 $\alpha$ -

methoxypregn-5-en-20-one (**2**) as reported<sup>2</sup> but with an improved yield (65%). In contrast, when the 20-ketone (**1**) was treated with the brominating agent as above but in the presence of pyridine (3 equiv.), instead of producing a higher yield of the methoxide (**2**) analogous to the results of Sollman and Dodson,<sup>5</sup> the 21-bromide (**3**) [m.p. 156–158 °C (lit.<sup>6</sup> m.p. 159–159.5 °C); <sup>1</sup>H n.m.r.  $\delta$  (CDCl<sub>3</sub>) 0.67 (3H, s, 3  $\times$  18-H), 1.62 (3H, s, 3  $\times$  19-H), 3.50 (1H, m, 3 $\alpha$ -H), 3.92 (2H, s, 2  $\times$  21-H), 5.38 (1H, m, 6-H);  $\nu_{\max}$  (KBr) 1719 cm<sup>-1</sup>; *m/z* 394 and 396 (*M*<sup>+</sup>)], was obtained regioselectively in 71% yield without the formation of the methoxide (**2**). This is the first direct bromination at C-21 of the 20-ketone (**1**) without affecting the integrity of the isolated double bond at C-5. This is a surprising result, in view of the usual reaction pathway of bromination with CuBr<sub>2</sub><sup>2</sup> or Br<sub>2</sub>,<sup>1</sup> and can be explained by the base functioning analogously to CaO<sup>7</sup> and probably promoting enolization, in the Hofmann sense, toward C-21.

The completion of a 21-hydroxy-20-one side-chain was accomplished by the use of an interesting reaction, the controlled alkaline hydrolysis<sup>4</sup> of the corresponding 21-bromo-20-one. When the bromoketone (**4**) [m.p. 184–185 °C (lit.<sup>8</sup> m.p.



**Scheme 1.** Reagents: i,  $\text{CuBr}_2$ , pyridine, MeOH; ii, 2.67 M  $\text{CrO}_3$ , acetone and then toluene-*p*-sulphonic acid, acetone; iii, NaOH and aqueous DMF or  $\text{K}_2\text{CO}_3$  and aqueous acetone.

190–191 °C);  $^1\text{H}$  n.m.r.  $\delta$  ( $\text{CDCl}_3$ ) 0.70 (3H, s,  $3 \times 18\text{-H}$ ), 1.18 (3H, s,  $3 \times 19\text{-H}$ ), 3.90 (2H, s,  $2 \times 21\text{-H}$ ), 5.73 (1H, s, 4-H),  $\nu_{\text{max}}$  (KBr) 1718, 1660  $\text{cm}^{-1}$ ], obtained by oxidation of the

bromide (3) with 2.67 M  $\text{CrO}_3$  and subsequent treatment with toluene-*p*-sulphonic acid, was treated with NaOH [1.2 equiv., 67% aqueous dimethylformamide (DMF),<sup>4e</sup> room temp., 30 min] or  $\text{K}_2\text{CO}_3$ <sup>4e</sup> (1 mol. equiv., 60% aqueous acetone, reflux, 1 h), deoxycorticosterone (5), m.p. 139–140 °C (lit.<sup>8</sup> 141–142 °C), was obtained in 85–95% yield. The ketol (5) was identical with the natural product in every respect and the overall yield, without purification and isolation of intermediates, was ca. 55%.

The obvious advantages of this sequence are that the 3 $\beta$ -hydroxy-5-ene system does not interfere with the reaction and a 21-bromo-20-ketone can be converted directly into the corresponding 21-hydroxide. 11-Deoxycortisol having a dihydroxyacetone side-chain has also been efficiently obtained (70%) by the controlled alkaline hydrolysis of 21-bromo-17 $\alpha$ -hydroxypregn-4-ene-3,20-dione.

We thank Dr. Toshio Nambara for mass spectra.

Received, 24th September 1982; Com. 1128

## References

- 1 For a review see E. P. Oliveto in 'Organic Reactions in Steroid Chemistry,' vol. II, eds. J. Fried and J. A. Edwards, Van Nostrand Reinhold, New York, 1972, pp. 127–217.
- 2 E. R. Glazier, *J. Org. Chem.*, 1962, **27**, 4397.
- 3 D. N. Kirk and M. P. Hartshorn in 'Steroid Reaction Mechanisms,' Elsevier, Amsterdam, 1968, p. 388.
- 4 (a) M. Numazawa and Y. Osawa, *J. Am. Chem. Soc.*, 1980, **102**, 5402; (b) M. Numazawa, K. Kimura, and M. Nagaoka, *Steroids*, 1981, **38**, 557; (c) M. Numazawa and M. Nagaoka, *Steroids*, 1982, **39**, 345; (d) M. Numazawa, M. Nagaoka, M. Tsuji, and Y. Osawa, *J. Chem. Soc., Chem. Commun.*, 1981, 383; (e) M. Numazawa, M. Nagaoka, and Y. Osawa, *J. Org. Chem.*, 1982, **47**, 4024.
- 5 P. B. Sollman and R. M. Dodson, *J. Org. Chem.*, 1961, **26**, 4180.
- 6 H. Reich and T. Reichstein, *Helv. Chim. Acta*, 1939, **22**, 1124.
- 7 H. J. Ringold and G. Stork, *J. Am. Chem. Soc.*, 1958, **80**, 250.
- 8 T. Reichstein, Swiss Patent No. 235,191, 1945 (*Chem. Abstr.*, 1949, **43**, 7056h).